Synthesis and Structure of a Phosphagermirene

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The first phosphagermirene has been prepared *via* the reaction of the phospha-alkyne Bu^tC=P with the germylene $\{(Me_3Si)_2CH\}_2Ge$.

Recently it has been demonstrated that phospha-alkynes (RC=P) will add carbene¹ and silylene² moieties to afford phosphirenes and phosphasilirenes, respectively. However, structural information is only available for the W(CO)₅ adducts of these interesting compounds. We report (i) synthesis of the first phosphagermirene, and (ii) the first X-ray structural data for this type of ring system.

In a typical reaction, $GeR_2 [R = (Me_3Si)_2CH]^3 (1.0 \text{ mmol})$ in n-hexane (50 ml) was treated with Bu⁴C=P⁴ (1.5 mmol) in 1,2-dimethoxyethane at 25 °C. The colour of the mixture changed immediately from orange to yellow. The solvent and excess of Bu⁴C=P were removed by evacuation and the residue was recrystallised from toluene to afford an 80% yield of yellow, crystalline 2,2-di[bis(trimethylsilyl)methyl]3-t-butylphosphagermirene (1) (m.p. 89-91 °C).

The ${}^{31}P$ n.m.r. chemical shift for (1)† falls in the region expected for P=C.⁵ Moreover, integration of the ${}^{1}H$ n.m.r.

signals reveals (a) a 2:1 ratio of $(Me_3Si)_2CH$ and Bu^t groups, and (b) that there are two environments for the Me_3Si groups.

The structure of (1) was confirmed by X-ray analysis (Figure 1).[‡] Note that there are two independent molecules per

$$Bu^{t}C \equiv P + R_{2}Ge \longrightarrow Bu^{t}C \equiv P$$

$$(1)$$

$$R = (Me_{3}Si)_{2}CH$$

‡ Crystal data for (1): C₁₉H₄₇GePSi₄, M = 491.49, monoclinic, $P2_1/n$ (No. 14), a = 23.074(5), b = 12.548(3), c = 23.235(5) Å, $\beta = 118.74(2)^\circ$, U = 5897.9 Å³, Z = 8, $D_c = 1.107$ g cm⁻³, μ (Mo- K_α) = 12.4 cm⁻¹; 7693 unique reflections measured with an Enraf-Nonius CAD-4 diffractometer over the range $3.0 \le 20 \le 45.0^\circ$ ($\theta/2\theta$ scan mode). Anisotropic decay and empirical absorption corrections were made. The structure was solved by direct methods and refined (difference Fourier; full matrix least-squares) using 4050 reflections with $I > 3.0\sigma$ (I). The final residuals were R = 0.0515 and $R_w = 0.0617$. Atomic co-ordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.

[†] N.m.r. spectra recorded at ambient temperature relative to 85% H_3PO_4 (³¹P) or Me₄Si (¹H and ¹³C): ³¹P{¹H} n.m.r. (n-hexane; 121.5 MHz) δ +315(s); ¹H n.m.r. (C₆D₆; 300 MHz) δ 1.43 (s, 9H, Bu¹), 1.2 (s, 18H, Me₃Si), and 1.7 (s, 18H, Me₃Si); ¹³C{¹H} n.m.r. (C₆D₆; 75.4 MHz) δ 3.0 (s, Me₃Si); 3.4 (s, Me₃Si), 10.8 [s, CH(SiMe₃)₂], 31.73 [d, ³J_{PC} 7.5 Hz, C(CH₃)₃], and 44.24 [d, ²J_{PC} 1.5 Hz, C(CH₃)₃].

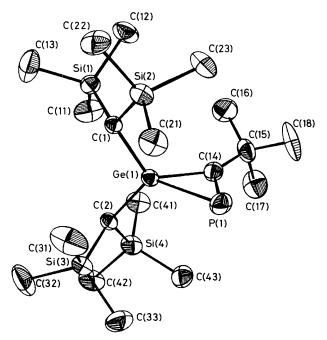


Figure 1. View (ORTEP) of (1) showing the atom numbering scheme for molecule 1. Important lengths (Å) and angles (°): molecule 1, Ge(1)-P(1) 2.298(4), Ge(1)-C(14) 1.932(11), P(1)-C(14) 1.661(11), P(1)-Ge(1)-C(14) 45.2(3), Ge(1)-P(1)-C(14) 55.7(4), Ge(1)-C(14)-P(1) 79.1(5); molecule 2, Ge(2)-P(2) 2.291(4), Ge(2)-C(24) 1.928(11), P(2)-C(24) 1.678(11), P(2)-Ge(2)-C(24) 45.9(3), Ge(2)-P(2) 78.6(4).

asymmetric unit. The average phosphorus–carbon bond distance [1.669(11) Å] is indicative of a bond order of two⁵; hence (1) may be considered to be isolobal⁶ with cyclopropene. The P–Ge distance [average 2.294(4)Å] is similar to those observed in the diphosphagermirane, $Et_2Ge(PBu^t)_2$ [2.301(2) and 2.300(2) Å].⁷ There are, however, significant differences in the intra-ring angles, the largest being P–C–Ge [average 78.8(5)°]. Finally, it is clear from the structure that the Me₃Si group non-equivalence results from diastereotopy.

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